

Direct Synthesis of Arylcarbamates from Amines, Alcohols and CO₂ over CeO₂ with 2-Cyanopyridine

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Abstract: We found that combination of CeO₂ and 2-cyanopyridine was effective for the direct conversion of CO₂ with aniline and methanol to methyl *N*-phenylcarbamate (MPC), which can be an important intermediate for the synthesis of polyurethane, and obtained high yield of MPC (97%) The catalyst system was applicable to the syntheses of various arylcarbamates from CO₂, alcohols and aniline derivatives, achieving high yields (up to >99%), which is far higher than those of the reported literatures.

Keywords: Cerium oxide, Carbo dioxide, Carbamate.

1. Introduction

Methyl *N*-phenylcarbamate (MPC) is a promising precursor for the synthesis of methylenediphenyl diisocyanate (MDI), which is an important intermediate of polyurethane. Conventionally, MPC is produced by using phosgene or CO, however, these reagents are hazardous and therefore greener methods are desired to be developed. Up to now, various alternative synthesis methods for MPC have been developed, and one-pot synthesis of MPC from aniline, methanol and CO₂ will be an eco-friendly method since water is an only byproduct, and cheap and abundant CO₂ can be used as a carbonyl source. However, it is difficult to obtain MPC in high yield due to the severe equilibrium limitation and low reactivity of aniline, and the reported yields of MPC was ≤10%. Therefore, development of effective methods for the direct MPC synthesis from CO₂, aniline and methanol is desirable. In our laboratory, we have investigated CeO₂-catalyzed CO₂ transformation to various valuable chemicals such as organic carbonates (1), ureas (2) and cyclic carbamate (3), and combination of CeO₂ and 2-cyanopyridine can work as an effective catalyst for the carbonate synthesis from CO₂ and alcohols and diols by removal of the produced H₂O via CeO₂-catalyzed hydration of 2-cyanopyridine to 2-picolinamide. Herein, we found that the combination of CeO₂ with 2-cyanopyridine was effective for the one-pot synthesis of MPC.

2. Experimental

CeO₂ was obtained by calcining CeO₂ (CeO₂-HS, Daiichi Kigenso Kagaku Kogyo Co., Ltd.) at 873 K for 3 h. Other metal oxides were also obtained by calcining commercial available ones or prepared ones at 873 or 773 K for 3 h. All the reactions were carried out in an autoclave reactor with inner volume of 190 ml. CeO₂ (1 mmol), aniline (5 mmol), methanol (75 mmol) and 2-cyanopyridine (75 mmol) were put into the autoclave together with a spinner, and then the reactor was purged with CO₂ (1 MPa) and pressurized with CO₂ (5 MPa) at room temperature. Then the reactor was heated to the reaction temperature (typically 403 K). After the reaction, the reactor was cooled down to room temperature, and the liquid products were collected by using methanol. Products in the liquid phases were analyzed by using GC-FID, GC/MS and HPLC.

3. Results and discussion

Direct MPC synthesis from aniline, methanol and CO₂ was carried out using various metal oxide and 2-cyanopyridine (Figure 1). Only 2-cyanopyridine or only CeO₂ hardly provided MPC, and *N*-phenyl-2-pyridinecarboxamidine, which is produced by the reaction of aniline and 2-cyanopyridine, was mainly obtained. In the case of CeO₂, SiO₂-Al₂O₃, Nb₂O₅, ZnO and SiO₂, MPC formation was observed, and among these metal oxides, CeO₂ provided the highest MPC selectivity with moderate conversion, and main other product was *N,N'*-diphenylurea (DPU), which can be produced from CO₂ and aniline. From these results,

combination of CeO₂ and 2-cyanopyridine was effective for the reaction. Time-course of MPC synthesis from aniline, methanol and CO₂ using CeO₂ and 2-cyanopyridine was studied. The selectivity of DPU was high (~75%) at low conversion level, and decreased with increasing the reaction time and that of MPC increased. This result suggests that DPU was an intermediate for MPC formation. In addition, theoretical amount of 2-picolinamide was observed, indicating that the produced H₂O was smoothly removed by the hydration of 2-cyanopyridine over CeO₂ catalyst, which will be responsible for high performance. At 8 h reaction time, MPC yield reached 97%, which is much higher than the reported MPC yields.

The catalyst system composed of CeO₂ and 2-cyanopyridine was applied to the arylcarbamate syntheses with various alcohols and aniline derivatives. Various substituted anilines were converted with methanol and CO₂ to the corresponding arylcarbamates in high yields, and moreover various primary and secondary alcohols were converted with aniline and CO₂ to the carbamates in high yields

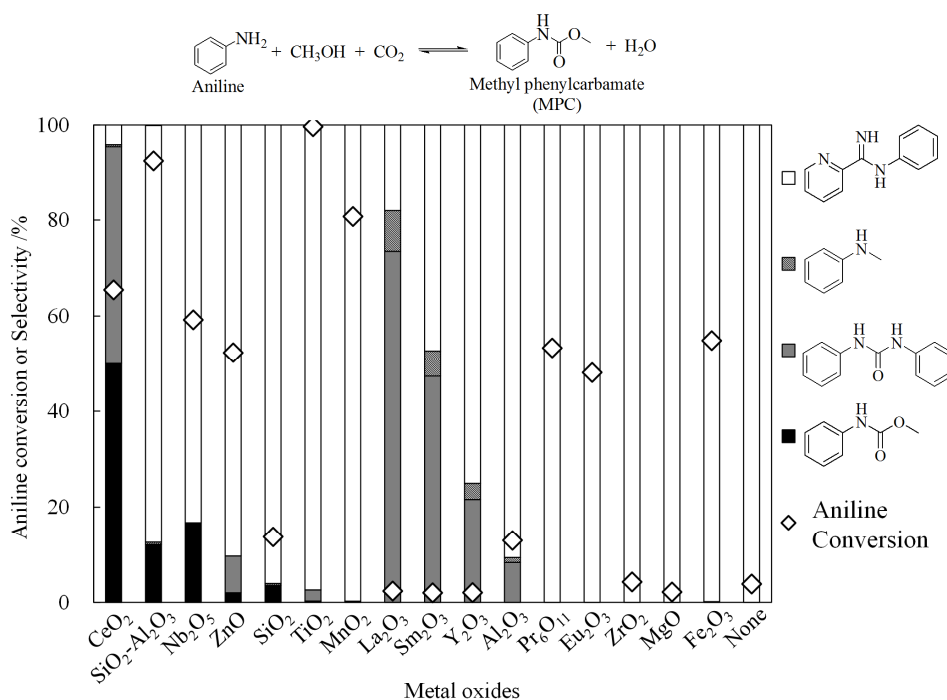


Figure 1. Catalyst screening for the synthesis of MPC from aniline, methanol and CO₂. Reaction conditions: metal oxides 0.34 g or 0 g, aniline 5.0 mmol, methanol 75 mmol, 2-cyanopyridine 75 mmol, CO₂ 5 MPa (at rt), 403 K, 6 h.

4. Conclusions

The combination catalyst system of CeO₂ and 2-cyanopyridine was effective for the direct synthesis of MPC from aniline, methanol and CO₂, achieving high MPC yield (97%). Smooth removal of the produced H₂O by hydration of 2-cyanopyridine over CeO₂ catalyst led to the high yield of MPC. The catalyst system was applicable to the synthesis of various arylcarbamates from CO₂, alcohols and aniline derivatives.

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