

Ni and Co promoted MoW alumina supported HDS catalysts prepared from mixed SiMo₃W₉ heteropolyacids

Mariia Nikulshina^{a,b*}, Alexander Mozhaev^a, Christine Lancelot^b, Maya Marinova^c, Pascal Blanchard^b, Edmond Payen^b, Carole Lamonier^b and Pavel Nikulshin^d

^aSamara State Technical University, Samara, 443100, Russia

^bUniv. Lille, CNRS, Centrale Lille, ENSCL, Univ. Artois, UMR 8181 - UCCS - Unité de Catalyse et Chimie du Solide, Lille, F-59000, France

^cInstitut Chevreul, University of Lille & CNRS, Villeneuve d'Ascq, F-59655, France

^dAll-Russian Research Institute of Oil Refining, Moscow, 111116, Russia

*Corresponding author: kulikova.m.s@yandex.ru

Abstract: Unpromoted and Ni/Co-promoted MoW HDS catalysts supported on alumina were prepared using mixed H₄SiMo₃W₉O₄₀ heteropolyacid and mixture of SiMo₁₂ and SiW₁₂ HPAs. All catalysts were characterized by N₂ physisorption, HRTEM, XPS and evaluated in hydrodesulfurization of dibenzothiophene (DBT) and naphthalene hydrogenation (HYD). It was found that both Ni/Co-MoW catalysts prepared from mixed HPA showed better performance in HDS of DBT and HYD of naphthalene than their counterparts prepared from two separate monometallic HPAs. Ni-promoted catalysts exhibited higher catalytic activity compared to Co-promoted ones.

Keywords: Hydrotreating, Heteropolyanion, NiMoWS.

1. Introduction

Developing highly active hydrodesulfurization (HDS) catalysts has been one of the most challenging and important subjects because of strengthening of environmental requirements for the sulfur content in fuels. Traditionally, bimetallic sulfide catalysts based on Mo or W, promoted by Ni or Co and supported on Al₂O₃ are used in the HDS processes. One of the approaches to improve catalytic activity can be the development of mixed MoW based catalysts. In this investigation, we report the synthesis and characterization of MoW catalysts promoted by Ni or Co prepared by using mixed H₄SiMo₃W₉O₄₀ heteropolyacid (HPA) of Keggin structure.

2. Experimental

Unpromoted and Ni/Co-promoted Mo(W) catalysts with the same surface density of metals - d(Mo+W) expressed by at.nm⁻² - were synthesized by incipient wetness method impregnation of the supports with aqueous solutions containing the required amounts of SiW₉Mo₃HPA, Ni or Co citrate with molar ratio Ni(Co)/(W+Mo) = 0.5. Moreover, the catalysts based on monometallic SiW₁₂HPA and mixture of SiMo₁₂ and SiW₁₂HPAs were prepared as references. The chemical composition of the synthesized catalysts is shown in Table 1.

Table 1. Chemical composition of prepared catalysts.

Catalyst	MoO ₃ , wt. %	WO ₃ , wt. %	NiO/CoO, wt. %	d(Mo+W), at.nm ⁻²
(Ni/Co)-W/Al		26.2	4.2	4.0 (4.2)*
(Ni/Co)-Mo ₃ W ₉ /Al	4.2	20.1	4.3	4.0 (4.2)
(Ni/Co)-(Mo+W)/Al	4.2	20.1	4.3	4.0 (4.2)

*density of metals d(Mo+W) calculated for Ni(Co)-promoted catalysts.

The hydrotreatment activities were measured on a model feed (1000 ppm of sulfur, 3 wt. % of naphthalene and toluene as a solvent) in a bench-scale flow reactor under pressure of hydrogen. Before testing, the catalysts were activated by sulphidation. A mixture of DMDS (2 wt. % of S) and decane at 3.5 MPa was used in a stepwise procedure conducted over 10 h at 240 °C and 8 h at 340 °C. The conditions of

the tests were the following: temperature 280 °C; pressure 3.0 MPa; feed space velocity 5 h⁻¹ (unpromoted samples) and 40 (promoted catalysts); a H₂ feedstock ratio 500 NL/L. The liquid product compositions of the samples collected every 0.5 h were determined using a GCMS-QP2010 Ultra. The HDS reaction was allowed to proceed for 10 h to obtain the steady-state activity of the catalysts. To evaluate the catalytic performances, rate constants of the DBT HDS and naphthalene HYD reactions were determined

3. Results and discussion

Figure 2 shows the hydrotreating results for the mixture of DBT and naphthalene over the prepared samples.

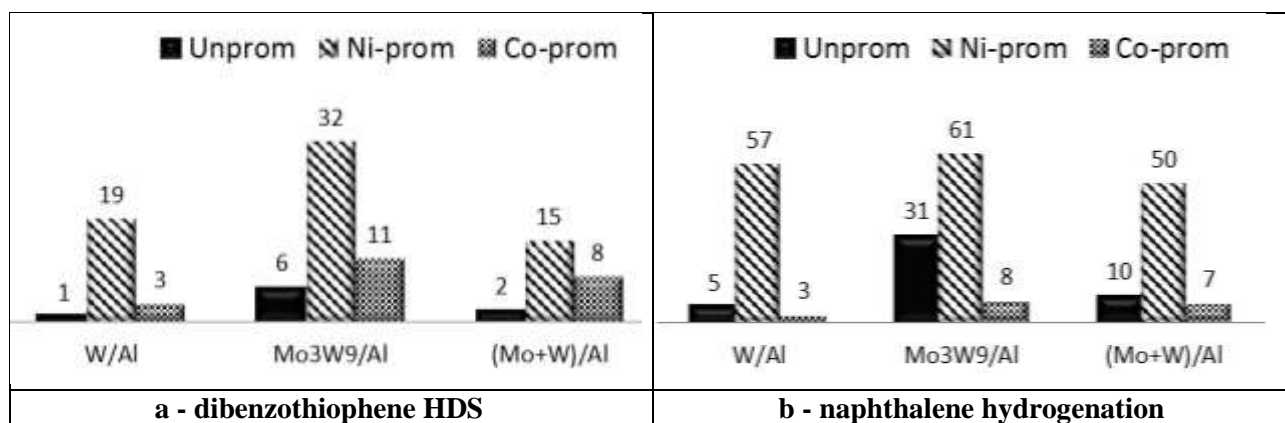


Figure 1. Dependence of the rate constants of the HDS DBT (a) and HYD naphthalene (b) reactions over Ni(Co)Mo(W)/Al catalysts on their chemical composition.

It was found, that the introduction of Mo from mixed precursor led to increasing HDS and HYD activities compared to that of SiW₁₂HPA based catalysts. As a result, all MoW containing reference samples prepared from two separate HPAs had lower catalytic activity than their counterpart based on mixed SiW₉Mo₃HPA. These results witness in favor of the creation of more active mixed Mo_xW_{1-x}S₂ active sites.

The formation of mixed Mo_xW_{1-x}S₂ slabs when mixed HPA was used for preparation of unpromoted catalyst was evidenced by EXAFS and HAADF analysis¹. Taking into account the above results, it can be assumed the creation of mixed Ni(Co)MoWS mixed active sites in case of promoted series. Then the low catalytic activity of Co-Mo₃W₉/Al compared to Ni-Mo₃W₉/Al is in good agreement with the reported DFT calculations² where P. Raybaud and co-workers predicted a synergetic effect for Ni promoted Mo_{0.1-x}W_xS₂ active phases relative NiMoS and NiWS ones and no synergetic effect for CoMo_{0.1-x}W_xS₂ phases with respect to CoMoS and CoWS ones.

4. Conclusions

It was found that the use of mixed SiMo₃W₉HPA as starting precursor yield high-efficient HDT catalysts. Mixed HPA based catalysts were more active in DBT HDS and naphthalene HYD reactions in contrast to their Mo+W references with the same surface density of the metals prepared from two separate monometallic HPAs.

References

1. M.Nikulshina, A. Mozhaev, C. Lancelot, M. Marinova, P. Blanchard, E. Payen, C. Lamonier, P. Nikulshin, Appl. Catal. B 224 (2018) 951.
2. C. Thomazeau, C. Geantet, M. Lacroix, M. Danot, V. Harle, P. Raybaud, Appl. Catal. A 322 (2007) 92.