

Drastic onset potential shift of Ta₃N₅ photoanode with Mg:GaN modification

Ela Nurlaela,^a Yutaka Sasaki,^a Mamiko Nakabayashi,^b Naoya Shibata,^b Taro Yamada,^a and Kazunari Domen^{*a}

[a] Dr. E. Nurlaela, Y. Sasaki, Dr. T. Yamada, Prof. K. Domen, School of Engineering, the University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-8656 (Japan)

Japan Technological Research Association of Artificial Photosynthetic Chemical Process 2-11-9 Iwamotocho, Chiyoda-ku, Tokyo 101-0032 (Japan)

E-mail: domen@chemsys.t.u-tokyo.ac.jp

[b] M. Nakabayashi, Prof. N. Shibata, Institute of Engineering Innovation, The University of Tokyo, 2-11-16 Yayoi, Bunkyo-ku, Tokyo 113-8656 (Japan)

Abstract: Ta₃N₅ is a promising material for photoelectrochemical (PEC) water splitting application thanks to its low bandgap energy and suitable band positions. An optimum current density and a long stability up to 12 h has been demonstrated. However, Ta₃N₅ suffers from high onset potential. Here we report a remarkable improvement of Ta₃N₅ onset potential with Mg:GaN modification on the surface followed by post annealing in N₂ or NH₃. PEC performances and characterization have revealed the mechanism of Mg activation/Ta₃N₅ damage recovery by N₂/NH₃ post annealing. Despite the current density loss, this finding unlocks the prospect of unassisted PEC water splitting.

Keywords: Ta₃N₅, photoanode, water splitting.

1. Introduction

Photoelectrochemical (PEC) water splitting offers an attractive solution to convert solar energy to hydrogen in the most eloquent ways in term of chemical robustness, high durability, and technological design for large scale application. The challenge lies on finding a suitable material for semiconductor electrodes which has to fulfill stringent requirements at once. Numerous materials have been explored as a suitable photoanode candidate and among them, Ta₃N₅ has emerged as one of the most attractive material due to its low bandgap energy (2.1 eV) and suitable band positions.¹⁻³ However, Ta₃N₅ suffers from poor stability and high onset potential.¹⁻³ Given that the flatband potential of Ta₃N₅ is more negative than water reduction potential (ca. 0 vs NHE), the reported onset potential is excessively positive.¹⁻³

In this work we report improvement of the onset potential of Ta₃N₅ photoanode by Mg:GaN modification followed by post annealing treatment. The sample exhibited a significant negative onset potential towards 0 V vs RHE, one of the best ever reported on Ta₃N₅ photoanode. The samples were fabricated by Mg:GaN deposition utilizing plasma-enhanced chemical vapor deposition (PCVD) on sputtered-Ta₃N₅.

2. Experimental

Ta₃N₅ thin films were prepared by reactive RF magnetron sputtering of TaO₅ followed by nitridation process. Mg:GaN was deposited on the surface of Ta₃N₅ by applying a plasma-enhanced chemical vapor deposition (PCVD) apparatus. Trimethylgallium as the Ga source and Bis(cyclopentadienyl)magnesium(II) as the Mg source were carried into the deposition chamber with N₂ flow (5 mL·min⁻¹). Radio frequency plasma was conducted at 1000 W with the additional N₂ flow (20 mL·min⁻¹). The Mg:GaN/Ta₃N₅ sample was then subjected to N₂ or NH₃ post annealing treatment. N₂ post annealing was conducted in different temperature from 600 – 800 °C for 30 min in the presence of 200 mL·min⁻¹ N₂ flow while NH₃ post annealing was done from 800 – 1000 °C for 1 h in the presence of 200 mL·min⁻¹ NH₃ flow.

The photoanode was fabricated by simply connecting the backside Ta substrate of the sample to a copper wire with indium solder and covered the unnecessary part with epoxy resin. The deposition of CoPi on Ta₃N₅ and Mg:GaN/Ta₃N₅ photoanodes was performed in a 0.1 M potassium phosphate (KPi) solution containing 0.5 mM Co(NO₃)₃ at pH 7 by chronoamperometry.

3. Results and discussion

The PEC performance of CoPi/Mg:GaN/Ta₃N₅ with N₂ or NH₃ post annealing was investigated. The linear sweep voltammetry (LSV) scans in Figure 1 show that upon Mg:GaN deposition followed by N₂ post annealing, the onset potential shifted toward 0 vs RHE. However, the current density decreased more than 20 folds. There was no negative onset potential shift upon NH₃ post annealing. However, the current density increased, reaching the original current density of Ta₃N₅.

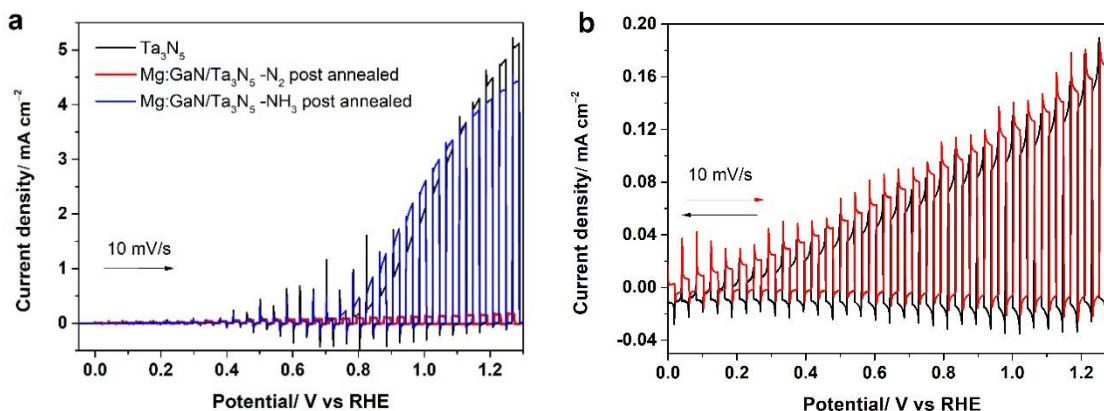


Figure 1. a) LSV of Ta₃N₅ and Mg:GaN/Ta₃N₅ after N₂ and NH₃ post annealing, b) Magnification of N₂ post annealed Mg:GaN/Ta₃N₅ LSV in Fig. a

Among all the characterization of Mg:GaN before and after N₂ post annealing, only XPS and Mott-Schottky detected some differences. It is apparent that Mg 1s peak is slightly shifted to more reduced state whereas the E_{fb} shifted to more negative value after N₂ post annealing. These findings are favorable conditions for the negative onset potential shift of Mg:GaN/Ta₃N₅. Mg:GaN/Ta₃N₅ possesses E_{VB} higher by 0.25 V than that of Ta₃N₅. This favors faster hole transfer from Ta₃N₅ to Mg:GaN. As for electrons, E_{fb} of Mg:GaN, higher by 0.1 V than E_{fb} of Ta₃N₅, facilitates effective electron transfer from Mg:GaN to Ta₃N₅. NH₃ post annealing was demonstrated to increase the current density of the sample without onset potential shift, which is probably due to accumulated effects evidenced from its characterization results. The surface damage of Ta₃N₅ was healed and the new pinholes appeared on the surface of Mg:GaN. Thus the PEC performance is solely coming from Ta₃N₅. The shift of E_{VB} to a more positive value seems to be an effect of the loss of Mg concentration, resembling undoped GaN.

4. Conclusions

In summary, a negative shift of onset potential has been realized on Mg:GaN modified-Ta₃N₅. Detailed studies revealed the correlation of the PEC performance with the results of structural and chemical characterization. N₂ post annealing treatment is a crucial step for further improvement of the onset potential reaching 0 vs RHE by Mg activation mechanism. The trade-off between low onset potential and current density decrease is due to the thermally induced damage on Ta₃N₅ near the surface during the process. Improvement and recovery of Ta₃N₅ photocurrent was realized by NH₃ post annealing however without negative shift of the onset potential.

References

1. W.J. Chun, A. Ishikawa, H. Fujisawa, T. Takata, JN Kondo, M. Hara, M. Kawai, Y. Matsumoto, K. Domen, *J. Phys. Chem. B* **2003**, *107*,1798-1803.
2. M. Li, W. Luo, D. Cao, X. Zhao, Z. Li, T. Yu, Z. Zou, *Angew. Chem. Int. Ed.* **2013**, *52*, 11016-11020.
3. G. Liu, S. Ye, P. Yan, F. Xiong, P. Fu, Z. Wang, Z. Chen, J. Shi, C. Li, *Energy Environ. Sci.* **2016**, *9*, 1327—1334.